

Kurzmitteilung / Short Communication

Synthesis of a Novel Annulated C₁-Bridged *ansa*-Metallocene System

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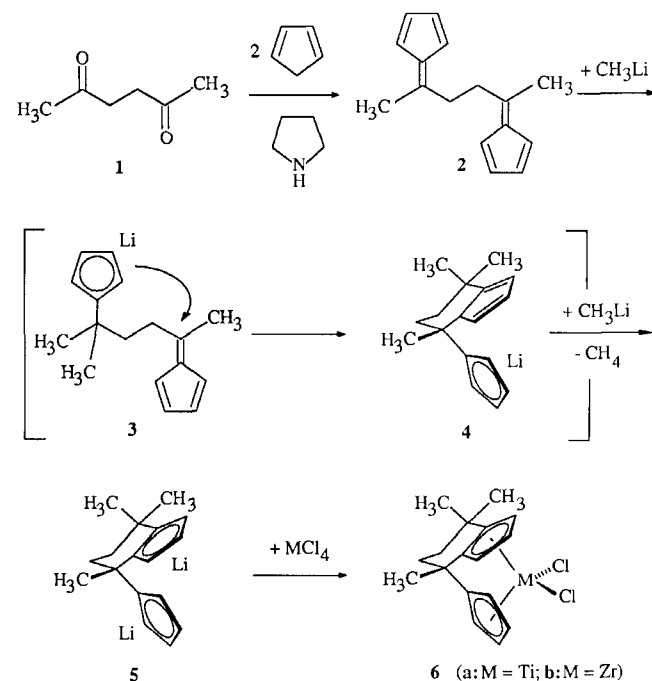
2,5-Hexanedione (**1**) was converted into the bis(fulvene) **2**, which was then treated with 2 molar equiv. of ethereal methylolithium to give the novel annulated C₁-bridged dilithium bis(cyclopentadienide) **5**. Reagent **5** was treated with MCl₄

(M = Ti, Zr) to yield the annulated C₁-bridged *ansa*-metallocene dichlorides **6a** and **6b**, respectively. The homogeneous **6a**, **6b**/methylalumoxane Ziegler-type systems were employed in propene polymerization reactions.

In the 1970s Sinn and Kaminsky discovered that the activation of group-4 bent metallocene complexes by excess oligomeric methylalumoxane generated extremely active homogeneous Ziegler-type catalysts for the polymerization of ethylene and some α -olefins^[1]. Then Brintzinger et al. disclosed their syntheses of chiral [CH₂]_n-bridged *ansa*-metallocenes of titanium and zirconium^[2]. The combination of these two discoveries has initiated a revolutionary development in stereoselective olefin polymerization^[3]. Since then *ansa*-metallocenes have been employed in some variety for the generation of homogeneous Ziegler-type catalysts, which are mostly derivatives or analogues of the original Brintzinger systems containing one- or two-carbon- or silicon-atom bridges^[4]. We have now synthesized a novel type of an *ansa*-metallocene system where the bridging hydrocarbyl moiety contains a fused annulated ring. The examples described below contain a single Cp–Cp bridging unit which constitutes an effective C₁ connection and at the same time is part of a C₄-*ansa* chain. This specific *ansa*-metallocene system, of which we have prepared the titanium and zirconium derivatives, is very rigid.

We have treated the bis(fulvene) **2**^[5] with two molar equivalents of methylolithium in ether at 0°C to obtain the dilithiated *ansa* ligand **5**. Presumably, the bis(fulvene) **2** is attacked by methylolithium at C-2 of the hydrocarbon chain^[6], thereby generating the substituted Cp anion **3** which then intramolecularly attacks the remaining fulvene functionality^[7] to yield product **4** with a six-membered ring. Subsequent deprotonation of **4** by the additional equivalent of methylolithium then leads to **5** which is treated with one molar equivalent of TiCl₄ in toluene (–78°C to ambient temp.) to give the metallocene **6a** (isolated in 13% yield after recrystallization from dichloromethane). The corresponding *ansa*-zirconocene dichloride **6b** is prepared analogously by treatment of **5** with zirconium tetrachloride (isolated in 14% yield).

The chiral zirconium complex **6b** (C₁ symmetry) exhibits seven cyclopentadienyl CH signals [¹H NMR, (CDCl₃)] at δ = 6.72, 6.69, 6.50, 6.43, 6.05, 5.86, and 5.72 [¹³C NMR (CDCl₃): δ = 125.9, 123.6, 117.9, 116.0, 110.3, 107.3, 105.8 plus three quaternary Cp signals at δ = 137.7, 121.3, and 113.7] and three methyl singlets (¹H NMR) at δ = 1.83, 1.33, and 1.10 [¹³C NMR: δ = 35.1, 27.7,



26.5). The titanium complex **6a** has been characterized by X-ray diffraction. In **6a** the titanium center is pseudotetrahedrally coordinated to two chloride ligands [Ti–Cl(1) 2.362(1), Ti–Cl(2) 2.331(1) Å] and two cyclopentadienyl groups. The Cp ligands are both η^5 -coordinated. The monosubstituted Cp ligand exhibits a pronounced variation of the metal–carbon bond lengths with short Ti–C(10)/C(11)/C(14) [2.336(2), 2.349(2), 2.351(2) Å] and longer Ti–C(12)/C(13) distances [2.446(3), 2.445(3) Å]. The C(12)–C(13) bond, which is located opposite to the substituted Cp carbon atom C(10), is shorter [1.382(4) Å] than the remaining C(sp²)–C(sp²) Cp bonds [1.411(3)–1.423(3) Å]. The disubstituted cyclopentadienyl ligand exhibits two short [Ti–C(1) 2.328(2), Ti–C(2) 2.291(2) Å] and three longer metal–carbon π contacts [Ti–C(3)/C(4)/C(5)

2.413(2), 2.482(2), 2.436(2) Å]. In this Cp ring the distal C(sp²)–C(sp²) linkages [C(3)–C(4) 1.391(3), C(2)–C(3) 1.411(3) Å] are only slightly shorter than the proximal C–C bonds [C(4)–C(5) 1.419(3), C(1)–C(5) 1.417(3) and C(1)–C(2) 1.418(3) Å].

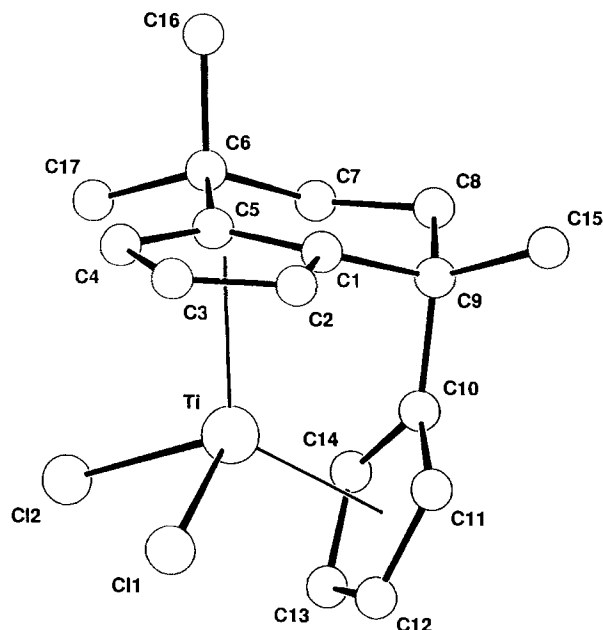


Figure 1. View of the molecular geometry of complex **6a** in the crystal

The annulated six-membered ring adopts a half-chair conformation. The ring carbon atom C(9) serves as the C₁ bridge between the two cyclopentadienyl rings. The C(1)–C(9) and C(10)–C(9) bond lengths [1.522(3), 1.533(3) Å] are in the typical C(sp²)–C(sp³) σ -bond range. The C(1)–C(9)–C(10) angle [97.1(2)°] deviates from tetrahedral. This is very typical of C₁-bridged *ansa*-titanocenes as is the out-of-plane bending of the C(1)–C(9)/C(10)–C(9) vectors toward the metal center by ca. 16–17°. In complex **6a** the Cp(centroid)–Ti–Cp(centroid) angle is 120.9° which is at the low end of the range of corresponding values (ca. 129°–121°) observed for *ansa*-titanocenes^[8]. The Cl(1)–Ti–Cl(2) angle in **6a** is 95.3(1)°.

Both complexes **6a** and **6b** have been used to generate active homogeneous Ziegler catalysts for propene polymerization. In a typical experiment the *ansa*-titanocene dichloride **6a** is treated with an excess of methylalumoxane ([Al]:[Ti] molar ratio ca. 550:1) in toluene solution at –50°C in the presence of propene. Partly isotactic high-molecular-weight polypropylene ($\bar{M}_n \approx 680000$) is formed (activity: $a = 300$ g polymer/g [Ti] · h). According to the ¹³C-NMR methyl pentade analysis^[9] only chain-end stereocontrol has taken place (ca. 40% mmmm, $\sigma = 0.78$). With increasing temperature the chiral metallocene backbone also takes part in the overall stereocontrol of the C–C coupling reaction. Similar to other previously described examples a situation is encountered here where both the “enantiomorphic site” and the chiral chain end are contributing simultaneously to the overall observed stereocontrol (double stereodifferentiation)^[6b,9b] [–40°C: $\bar{M}_n = 290000$, 40% mmmm, $\omega = 0.29$, $\alpha = 0.88$, $\sigma = 0.77$; –30°C: $\bar{M}_n = 215000$, $a = 640$, 22% mmmm, $\omega = 0.39$, $\alpha = 0.88$, $\sigma = 0.66$; –20°C: $\bar{M}_n = 56000$, 24% mmmm, $\omega = 0.56$, $\alpha = 0.78$, $\sigma = 0.66$; –11°C: $\bar{M}_n = 23000$, $a = 1100$, 22% mmmm, $\omega = 0.45$, $\alpha = 0.77$, $\sigma = 0.66$].

Activation of the analogous *ansa*-zirconocene dichloride **6b** with methylalumoxane ([Al]:[Zr] = 310:1) generates a catalyst which produces a polypropylene with a different stereochemical characteristic. At –30°C a polymer is obtained ($a = 330$; $\bar{M}_n = 26000$)

that is slightly syndiotactic. In the methyl region of the ¹³C-NMR spectrum (1,2,4-trichlorobenzene, +90°C) the mmmm signal is absent (<2% mmmm, 21% rrrr, 20% mrrr; chain-end control model: $\sigma = 0.32$). At –50°C ($a = 17$; $\bar{M}_n = 29000$, 23% rrrr, 20% mrrr, $\sigma = 0.31$) and –5°C ($a = 1500$; $\bar{M}_n = 5000$, 18% rrrr, 19% mrrr, $\sigma = 0.35$) similar polypropylenes are formed.

From the structure of the titanium compound it appears that the group-4 metal centers in the new *ansa*-metallocenes **6** and the catalysts derived thereof are rather exposed and easy to approach by incoming ligands. This results in high catalyst activities and a rapid decrease of polymer molecular weights with increasing temperatures. This special feature may provide a very suitable basis for developing novel homogeneous Ziegler-type systems to be employed in catalytic carbon–carbon coupling processes for the synthesis of monomeric organic products. We have begun to prepare derivatives of **6** that can be used as active catalysts in organic synthesis.

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Experimental

All reactions were carried out under Ar using Schlenk-type glassware or in a glovebox. Solvents were dried and distilled under Ar prior to use. For additional general information including a list of spectrometers used see ref.^[9b].

Preparation of 6a: The lithium reagent **5** · 2 Et₂O was prepared by addition of 62.4 ml of a 1.6 M ethereal CH₃Li solution to a solution of 10.0 g (47.5 mmol) of bis(fulvene) **2**^[5] in 150 ml of ether; 13.2 g (89%) of **5** was isolated after washing with pentane. Complex **6a** was obtained from the reaction of TiCl₄ (3.3 g, 17.5 mmol) with 5.5 g (17.5 mmol) of **5** · 2 Et₂O in 200 ml of toluene (–78°C to room temp.); yield 780 mg (13%), m.p. 260°C (DSC). – ¹H NMR (CDCl₃): $\delta = 7.05$, 6.92, 6.80, 6.78, 5.86, 5.80, 5.60 (m, each 1H, cyclopentadienyl CH), 2.5–2.2 (m, 2H, CH₂), 1.99–1.82 and 1.75–1.63 (m, 2H, CH₂), 1.86, 1.25, 1.11 (s, each 3H, CH₃). – ¹³C NMR (CDCl₃): $\delta = 142.3$, 114.8, 106.9, 37.0, 33.7 (quat. C), 134.1, 131.6, 126.3, 126.2, 113.6, 112.1, 111.1 (CH), 35.3, 31.5 (CH₂), 35.4, 27.4, 25.5 (CH₃). – C₁₇H₂₀Cl₂Ti (343.15): calcd. C 59.50, H 5.87; found C 59.54, H 5.97. – X-ray crystal structure analysis: monoclinic; space group *P*₂₁/*n*; $a = 9.233(2)$, $b = 10.543(1)$, $c = 16.665(1)$ Å; $\beta = 105.62(1)^\circ$; $Z = 4$; $\lambda = 0.71069$ Å; 3889 total, 3565 independent, 2914 observed reflections; 181 refined parameters; $R = 0.033$, $R_w = 0.042$ ^[10].

Preparation of 6b: Complex **6b** was prepared by reaction of 9.6 g (30.7 mmol) of **5** with 7.5 g (32.2 mmol) of ZrCl₄ analogously as described above; yield 1.7 g (14%) after extraction with pentane, m.p. 223°C (decomp.; DSC). – C₁₇H₂₀Cl₂Zr (386.47): calcd. C 52.83, H 5.22; found C 52.35, H 5.19. – For spectral data see text.

[1] H. Sinn, W. Kaminsky, *Adv. Organomet. Chem.* **1980**, *18*, 99, and references cited therein.

[2] H. Schnutenhaus, H. H. Brintzinger, *Angew. Chem.* **1979**, *91*, 837; *Angew. Chem. Int. Ed. Engl.* **1979**, *18*, 777.

[3] W. Kaminsky, K. K lper, H. H. Brintzinger, F. R. W. P. Wild, *Angew. Chem.* **1985**, *97*, 507; *Angew. Chem. Int. Ed. Engl.* **1985**, *24*, 507; J. A. Ewen, *J. Am. Chem. Soc.* **1984**, *106*, 6355.

[4] See e.g.: P. Burger, J. Diebold, S. Gutmann, H.-U. Hund, H. H. Brintzinger, *Organometallics* **1992**, *11*, 1319; B. Dorer, J. Diebold, O. Weynand, H. H. Brintzinger, *J. Organomet. Chem.* **1992**, *427*, 245; J. A. Ewen, L. Haspeslagh, J. L. Atwood, H. Zhang, *J. Am. Chem. Soc.* **1987**, *109*, 6544; C. M. Fendrick, L. D. Schertz, V. W. Day, T. J. Marks, *Organometallics* **1988**, *7*, 1828; W. A. Herrmann, J. Rohrmann, E. Herdtweck, W. Spa-

- leck, A. Winter, *Angew. Chem.* **1989**, *101*, 1536; *Angew. Chem. Int. Ed. Engl.* **1989**, *28*, 1511; W. Spaleck, M. Antberg, J. Rohrmann, A. Winter, B. Bachmann, P. Kiprof, J. Behm, W. A. Herrmann, *Angew. Chem.* **1992**, *104*, 1373; *Angew. Chem. Int. Ed. Engl.* **1992**, *31*, 1347; S. Collins, Y. Hong, R. Ramanachandran, N. J. Taylor, *Organometallics* **1991**, *10*, 2349; D. T. Mallin, M. D. Rausch, Y.-G. Lin, S. Dong, J. C. W. Chien, *J. Am. Chem. Soc.* **1990**, *112*, 2030; G. H. Llinas, S.-H. Dong, D. T. Mallin, M. D. Rausch, Y.-G. Lin, H. H. Winter, J. C. W. Chien, *Macromolecules* **1992**, *25*, 1242; J. A. Bandy, M. L. H. Green, I. M. Gardiner, K. Prout, *J. Chem. Soc., Dalton Trans.* **1991**, 2207; R. Gomez, T. Cuenca, P. Royo, E. Hovestreydt, *Organometallics* **1991**, *10*, 2516; G. Erker, S. Wilker, C. Krüger, R. Goddard, *J. Am. Chem. Soc.* **1992**, *114*, 10983; G. Erker, S. Wilker, C. Krüger, M. Nolte, *Organometallics* **1993**, *12*, 2140; and literature cited in these references.
- [5] M. S. Erickson, J. M. Cronan, J. G. Garcia, M. L. Mc Laughlin, *J. Org. Chem.* **1992**, *57*, 2504; for the general synthetic method used see: K. J. Stone, R. D. Little, *J. Org. Chem.* **1984**, *49*, 1849.
- [6] [6a] P. Renaut, G. Tainturier, B. Gautheron, *J. Organomet. Chem.* **1978**, *148*, 35, 43. — [6b] G. Erker, R. Nolte, Y.-H. Tsay, C. Krüger, *Angew. Chem.* **1989**, *101*, 642; *Angew. Chem. Int. Ed. Engl.* **1989**, *28*, 628.
- [7] For related coupling reactions involving fulvenes see e.g.: K. Hafner, *Angew. Chem.* **1958**, *70*, 419; M. Neuenschwander, P. Kronig, S. Schönholzer, M. Slongo, B. Uebersax, C. Rentsch, *Croat. Chem. Acta* **1980**, *53*, 625; P. Kronig, M. Slongo, M. Neuenschwander, *Makromol. Chem.* **1982**, *163*, 359.
- [8] J. A. Smith, J. von Seyerl, G. Huttner, H. H. Brintzinger, *J. Organomet. Chem.* **1979**, *173*, 175; W. Röhl, L. Zsolnai, G. Huttner, H. H. Brintzinger, *J. Organomet. Chem.* **1987**, *322*, 65; I. E. Nifant'ev, A. V. Churakov, I. F. Urazowski, Sh. G. Mkoyan, L. O. Atovmyan, *J. Organomet. Chem.* **1992**, *435*, 37; T. K. Hollis, A. L. Rheingold, N. P. Robinson, J. Whelan, B. Bosnich, *Organometallics* **1992**, *11*, 2812; M. E. Huttenloch, J. Diebold, U. Rief, H.-H. Brintzinger, A. M. Gilbert, T. J. Katz, *Organometallics* **1992**, *11*, 3600; Z. Chen, R. L. Halterman, *J. Am. Chem. Soc.* **1992**, *114*, 2276.
- [9] [9a] F. A. Bovey, G. V. D. Tiers, *J. Polym. Sci.* **1960**, *44*, 173; R. A. Sheldon, T. Fueno, T. Tsuntsuga, J. Kurukawa, *J. Polym. Sci., Part B: Polym. Lett.* **1965**, *3*, 23; J. Inoue, Y. Itabashi, R. Chujo, Y. Doi, *Polymer* **1984**, *25*, 1640. — [9b] G. Erker, R. Nolte, R. Aul, S. Wilker, C. Krüger, R. Noe, *J. Am. Chem. Soc.* **1991**, *113*, 7594.
- [10] Further details of the crystal structure investigation are available on request from the Fachinformationszentrum Karlsruhe, Gesellschaft für wissenschaftlich-technische Information mbH, D-76344 Eggenstein-Leopoldshafen, Germany, on quoting the depository number CSD-58408, the names of the authors, and the journal citation.

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